

# Physics Basic Formula's for IIT & JEE

## Work Function

The minimal amount of energy needed to remove one electron from the surface of a particular solid to infinity is the work function, which is a property of a material.

$$W = hv_0 = \frac{hc}{\lambda}$$

## Intensity in terms of electric field

A unit charge placed at a given area would experience that point's electric field intensity as a force. The intensity of the electric field is a vector quantity in nature.

$$I = \frac{1}{2} \epsilon_0 E^2 \cdot c$$

## Momentum of one photon

$$\text{Momentum} = \frac{h}{\lambda}$$

## Energy

$$\Delta E = \frac{12400}{\lambda(\text{\AA})} \text{ eV}$$

## De Broglie Wavelength

The de Broglie wavelength is a crucial topic to understand while studying quantum mechanics. De Broglie wavelength is a term used to describe the wavelength ( $\lambda$ ) that is connected to an item in relation to its momentum and mass.

$$\lambda = \frac{h}{mv} = \frac{h}{p} = \frac{1}{\sqrt{2mKE}}$$

## Radius and speed of electron in hydrogen like atoms

$$r_n = \frac{n^2}{Z} a_0 \text{ Where } a_0 = 0.529 \text{\AA}$$

$$V_n = \frac{Z}{n} V_0 \text{ where } V_0 = 2.19 \times 10^8 \frac{m}{s}$$

## Energy in nth orbit

$$E_n = E_1 \cdot \frac{Z^2}{n^2} \text{ Where } E_1 = -13.6 \text{ eV}$$

## Wavelength corresponding to spectral lines

$$\frac{1}{\lambda} = R \left[ \frac{1}{n_1^2} - \frac{1}{n_2^2} \right]$$

For Lyman series  $n_1 = 1$  and  $n_2 = 2, 3, 4, ..$

For Balmer Series  $n_1 = 2$  and  $n_2 = 3, 4, 5, ...$

For Paschen series  $n_1 = 3$  and  $n_2 = 4, 5, 6, ...$

### Total number of possible transition for nth state

$$\frac{n(n-1)}{2}$$

### Minimum wavelength for X-rays

$$\lambda = \frac{hc}{eV_0} = \frac{12400}{V_0(\text{Volt})} \text{ \AA}$$

### Moseley's Law

$$\sqrt{\bar{V}} = a(z - b)$$

### Average radius of nucleus is written as

$$R = R_0 A^{\frac{1}{3}} \text{ where } R_0 = 1.1 \times 10^{-15} \text{ M}$$

A is the mass

### Activity of a sample

$$A = A_0 e^{-\lambda t}$$

### Half life

$$T_{\frac{1}{2}} = \frac{0.693}{\lambda}$$

### Average Life

$$T_{avg} = \frac{T_{\frac{1}{2}}}{0.693}$$

A radioactive nucleus can decay by two different processes having half lives  $t_1$  and  $t_2$  respectively. Effective half-life of nucleus is given by

$$\frac{1}{t} = \frac{1}{t_1} + \frac{1}{t_2}$$

### If effect of nucleus motion is considered

$$r_n = (0.529 \text{ \AA}) \frac{n_2}{Z} \cdot \frac{m}{\mu}$$

$$E_n = (-13.6 \text{ eV}) \frac{Z^2}{n_2} \cdot \frac{\mu}{m}$$

Here  $\mu$  is reduced mass

$$\mu = \frac{Mm}{(M+m)}, \text{ Where } M \text{ is Mass of nucleus}$$